

Study of the hepatoprotective effect of *Juniperus phoenicea* constituents

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Abstract: Different fraction obtained from the aerial parts of *Juniperus phoenicea* showed significant activity as hepatoprotective when investigated against carbon tetrachloride induced liver injury. The hepatoprotective activity was evaluated through the quantification of biochemical parameters and confirmed using histopathology study. Phytochemical investigation of the petroleum ether, chloroform and methanol fractions utilizing different chromatographic techniques resulted in the isolation of five known diterpenoids namely: 13-epicupressic acid (**1**), imbricatolic acid (**2**), 7 α -hydroxysandaracopimaric acid (**3**), 3 β -hydroxysandaracopimaric acid (**4**), isopimaric acid (**5**), four flavonoid derivatives: cupressuflavone (**6**), hinokiflavone (**7**), hypolaetin-7-O- β -xylopyranoside (**9**), (-) catechin (**10**), in addition to sucrose (**8**). Both physical and spectral data were used for structure determination and all isolates were evaluated for their hepatoprotective activity. Compounds **2** and **6** were effective, however; **7** was the most active. Hepatoprotective activity of **7** is comparable with the standard drug silymarin in reducing the elevated liver enzymes and restoring normal appearance of hepatocytes. Hepatoprotective effect of combination of **6**, **7** and silymarin with the diterpene sugiol was also explored.

Keywords: *Juniperus phoenicea*; diterpenoids; flavonoids; Hepatoprotective; combined effect.

INTRODUCTION

Juniperus phoenicea is a tree or shrub up to 8 meters tall known locally as "Arayar". In Saudi Arabia, this species extends from the Mediterranean region southwards to as far south as Taif (Chaudhary, 1999). Previous phytochemical studies of *J. phoenicea* revealed that diterpenes are the main secondary metabolites of the plant (Muhammad *et al.*, 1992; Mossa *et al.*, 2004; Muhammad *et al.*, 1995; Muhammad *et al.*, 1996). The diterpene sandaracopimaric acid showed a lipoxxygenase inhibition effect (Muhammad *et al.*, 1996). Several biflavanoids including amentoflavone, cupressuflavone, robustaflavone, hinokiflavone and mono-O-methylhinokiflavone were isolated from the leaf extract of *J. phoenicea* (Fatma *et al.*, 1979). Phenylpropanoids (Hussein *et al.*, 2003; Comte *et al.*, 1996c; Comte *et al.*, 1997), flavonoids (Maatooq *et al.*, 1998) and furanones (Comte *et al.*, 1996a; Comte *et al.*, 1996b) were also isolated from the plant. Recently, amentoflavone, cupressuflavone, myricitrin, quercetin, cosmosin, quercitrin, *p*-coumaric acid and caffeic acid were isolated from the leaves of *J. phoenicea* growing in Egypt (Ali *et al.*, 2010). Deoxydopodophyllotoxin and β -peltatin-A methyl ether were identified as the major cytotoxic components of *J. phoenicea* (Cairnes *et al.*, 1980). The aqueous decoction of the berries showed antihepatotoxic effect (Aboul-Ela *et al.*, 2005). The ethanol extract and fractions obtained by liquid-liquid fractionation of the aerial parts showed

promising hepatoprotective effect (Alqasoumi *et al.*, 2009).

MATERIALS AND METHODS

General

Melting points were determined in open capillary tubes using *Thermosystem FP800 Mettler FP80* central processor supplied with *FP81 MBC* cell apparatus, and were uncorrected. Ultraviolet absorption spectra were obtained in methanol and with different shift reagents on a *Unicum Heyios a* UV-Visible spectrophotometer. ¹H and ¹³C NMR spectra were recorded on a *Bruker DRX-500* (Central Lab at the College of Pharmacy, King Saud University) spectrometer operating at 500 MHz for proton and 125 MHz for carbon, respectively. The chemical shift values are reported in δ (ppm) relative to the internal standard TMS or residual solvent peak, the coupling constants (*J*) are reported in Hertz (Hz). 2D-NMR experiments (COSY, HSQC, HMBC and NOESY) were obtained using standard *Bruker* program. MS were obtained using Liquid Chromatography/Mass Spectrometer (*Quattro micro API*) equipped with a Z-spray electrospray ion source (*Micromass*[®], *Quattro micro*[™], *WATERS*) and Gas Chromatography/Mass Spectrometer (6890N GC/5973 *Inert MS*; Agilent Technologies). Centrifugal preparative TLC (CPTLC) was performed using Chromatotron (Harrison Research Inc. model 7924): 1 mm silica gel P254 disc. Non UV active components were detected on TLC plates using

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vanillin/H₂SO₄ spray reagent followed by heat at 105°C for 5 minutes.

Plant materials

The aerial parts (twigs, leaves, fruits) of *Juniperus phoenicea* (L.) sub. sp. *phoenicea* (Cupressaceae) were collected at 15km south of Al Hadda, near Taif, Saudi Arabia, in January 2004, 2009. The plants were identified by Dr. Mohammad Atiqur Rahman, taxonomist of the Medicinal, Aromatic and Poisonous Plants Research Center (MAPPRC), College of Pharmacy, King Saud University, Riyadh, Saudi Arabia. A voucher specimen (# 14166) was deposited at the herbarium of this center.

Extraction, fractionation and purification

The dried ground aerial parts (2600g) were extracted to exhaustion by percolation at room temperature with 90% ethanol (16L), and the extract was evaporated *in vacuo* to leave 577g of residue. A portion of this residue (502g) was dissolved in 1200ml of 20% aqueous methanol and defatted with petroleum ether (500ml×3) to yield 174.82g petroleum ether soluble fraction. The aqueous methanol fraction was diluted with water until a 40% aqueous methanol mixture was produced and this was partitioned with chloroform (500ml×3) to yield 72.83g of chloroform soluble fraction and 254.35g of aqueous methanol soluble fraction. Preliminary hepatoprotective screening of the crude extract and all fractions (table 34 and 35) showed that the hepatoprotective activity of the ethanol extract was distributed between all phases.

Petroleum ether fraction

Ten grams of the petroleum ether extract were fractionated using vacuum liquid chromatography (VLC) technique on silica gel (15×10 cm i.d., 250g) eluting with hexane containing 10% increments of ethyl acetate to yield seven fractions. Fraction 1 (1.2/g) was further purified using CPTLC (4/mm silica gel GF₂₅₄ disk, solvent: 15% EtOAc in hexane) to give 230/mg of **1** as amorphous powder ($R_f=0.39$, solvent system: 30% EtOAc in hexane) and **2** as an oily residue ($R_f=0.48$, solvent system: 50% EtOAc in hexane). Preparative TLC of fraction 2 (130 mg) on C₁₈ TLC plates using 20% H₂O in CH₃CN as developing system resulted in the isolation of 17/mg of **3** as amorphous powder. Fraction 3 (0.5g) was rechromatographed on silica gel CC (30×1.5 cm i.d., 35g) eluting with 10% MeOH in CHCl₃ to give seven fractions A-G after TLC screening. Further purification of fractions E and F by Chromatotron (1mm silica gel GF₂₅₄ disk, solvent: 5% MeOH in CHCl₃) led to the isolation of compound **4** (67/mg) as amorphous powder ($R_f=0.36$; solvent system: 10% MeOH in CHCl₃).

Chloroform fraction

Part of the chloroform soluble fraction (67.83g) was fractionated using Sephadex LH-20 column (100 × 5 cm i.d., 700g). The column was eluted with the following

sequence of solvents mixture using CH₂Cl₂-Me₂CO (4:1, 2 L), CH₂Cl₂-Me₂CO (3:2, 2 L), CH₂Cl₂-MeOH (1:4, 1 L) and MeOH (2 L) to afford, in order of elution, three fractions [1 (20.41g), 2 (36.05g) and 3 (11.37g)]. The hepatoprotective assay showed that the activity was apportioned between fraction 2 (6.5g) and 3 (15.4g).

Fraction 2 (3.4 g) was subjected to a silica gel column (80 × 3 cm i.d., 105 g), eluted with CHCl₃ and Me₂CO with increasing concentration of Me₂CO [CHCl₃; CHCl₃-Me₂CO /9:1, 8:2, 7:3, 6:4, and Me₂CO]; fractions 250 ml each were collected, screened by TLC (solvent system: 50% EtOAc in hexane) and similar fractions were pooled to give 3 fractions. Fraction A (1.3g) was subsequently purified by CPTLC (4mm silica gel GF₂₅₄ disk, solvent: 20% EtOAc in hexane) to give 90mg of **1** ($R_f = 0.69$, solvent system: 40% EtOAc in hexane) as amorphous powder. Fraction B (1.5g) was rechromatographed on LiChroprep[®] RP-18 (25-40µm, MERCK) column (60 × 2.5 cm i.d., 60g) using 100% CH₃CN as eluant to give 55mg of **5** as amorphous powder ($R_f= 0.37$, solvent system: 100% CH₃CN, RP-18 plates).

Portion of fraction 3 (6.5 g) was rechromatographed on a Sephadex LH-20 column (80×4 cm i.d., 200 g) and eluted with MeOH-H₂O (3:1, 2L) to yield **6** and **7** as amorphous powders ($R_f= 0.36$ and $R_f= 0.50$ respectively, solvent system: 10% MeOH in CHCl₃ containing 0.1% HCO₂H).

Methanol fraction

The methanol fraction (249.35g) was fractionated by chromatography on Sephadex LH-20 (100×5 cm i.d., 700g) using the solvents CH₂Cl₂-MeOH (4:1, 2/L), CH₂Cl₂-MeOH (3:2, 2/L), CH₂Cl₂-MeOH (1:4, 1/L) and MeOH (2/L) to yield four fractions [1 (12.9g), 2 (125.95g), 3 (60.4g) and 4 (50.1g)]. The hepatoprotective assay indicated that the activity was mainly distributed between fraction 2 (125.95g) and 3 (60.4g); see table 41.

A sample of fraction 2 (5g) was subjected to silica gel CC (90×3 cm i.d., 130g) eluted with CHCl₃ followed by mixtures of CHCl₃ and MeOH. Polarity was increased by increasing the MeOH contents to obtain 7 fractions. The fractions 2-4 (1.6g) were subjected to a Sephadex LH-20 CC (60×3 cm i.d., 90g) eluting with CH₂Cl₂-MeOH (4:1, 500 ml) and CH₂Cl₂-MeOH (3:2, 500ml). Fractions eluted with CH₂Cl₂-MeOH (3:2, 500/ml) were further purified over LiChroprep[®] RP-18 CC (60×2.5/cm i.d., 30g) eluting with 18% CH₃CN-H₂O to provide compound **8** (356/mg) as white crystals.

Fraction 3 (25 g) was dissolved with H₂O and partitioned with EtOAc affording an EtOAc soluble fraction (2.7 g) and a water soluble fraction (22.3 g). The EtOAc-soluble fraction was subjected to silica gel CC (60×4 cm i.d., 90g) using a CHCl₃-MeOH gradient (from 19:1 to 0:1 v/v); fractions were collected, screened by TLC (solvent

system: 25% MeOH in CHCl₃) and pooled into two fractions. Fraction A (800mg) was further purified on a Sephadex LH 20 column (60×3 cm i.d., 60g) using MeOH as eluant followed by crystallization from MeOH to give **9** (49/mg). Fraction B (1.2g) was first subjected to silica gel CC (60×1.5 cm i.d., 40g) eluted with CHCl₃-MeOH (from 18:2 to 14:6 v/v), the semi pure fractions were further purified over Sephadex LH 20 column (60×3 cm i.d., 60g) using MeOH as eluant to yield pure **10** (175 mg).

13-Epicupressic acid (1). M.P. 104-105 °C. $[\alpha]_D = +22.6$ ($c = 0.031$, CHCl₃). IR ν_{\max} (KBr): 3510, 3090, 1720, 1645, 1250, 1225, 1150, 990, 925 and 875 cm⁻¹. ¹H- and ¹³C-NMR (CDCl₃): tables 1 and 2. ESIMS: 343 (100, [M + Na]⁺), calc. for C₂₀H₃₂O₃.

Imbricatolic acid (2). $[\alpha]_D = +45.1$ ($c = 0.032$, CHCl₃). IR ν_{\max} (film): 3500, 3000, 2934, 2873, 1740, 1694, 1645, 1240 and 890 cm⁻¹. ¹H- and ¹³C-NMR (CDCl₃): Tables 1 and 2. ESIMS: 345 (100, [M + Na]⁺), calc. for C₂₀H₃₄O₃.

7 α -Hydroxysandaracopimaric acid (3). M.P. 132-134 °C. $[\alpha]_D = -80.0$ ($c = 0.023$, MeOH); IR ν_{\max} (KBr): 3400, 3070, 1720, 1640, 1245, 995 and 905 cm⁻¹. ¹H- and ¹³C-NMR (CD₃OD): Tables 1 and 2. ESIMS: 341 (100, [M + Na]⁺), 357 (45, [M + K]⁺), calc. for C₂₀H₃₀O₃.

β -Hydroxysandaracopimaric acid (4): M.P. 249-251°C. $[\alpha]_D = +29.5$ ($c = 0.025$, MeOH). IR ν_{\max} (KBr): 3495, 3280, 2550, 1696, 1085, 988, 910 and 858 cm⁻¹; ESIMS (m/z 341 [M + Na]⁺ and 357 9100, [M + K]⁺), calc. for C₂₀H₃₀O₃.

Isopimaric acid (5): M.P. 132-134 °C. $[\alpha]_D = -14.8$ ($c = 0.025$, CHCl₃). IR ν_{\max} (KBr): 3300-2500, 1690 (COOH), 3090, 1640, 1000, 910 and 900 cm⁻¹; ¹H- and ¹³C-NMR (CD₃OD): Tables 1 and 2. ESIMS: 325 (100, [M + Na]⁺), 341(36, [M + K]⁺), calc. for C₂₀H₃₀O₂.

Cupressuflavone (6). M.P. 392 °C Decomp. UV λ_{\max} (MeOH) 329, 276 and 226 nm. IR ν_{\max} (KBr): 3440, 1640, 1610, 1580 and 1165 cm⁻¹. ¹H- and ¹³C-NMR (DMSO-*d*₆): Tables 3 and 4. ESIMS: 539 (100, [M + 1]⁺), 561 (74, [M+Na]⁺), 577 (32, [M+K]⁺), calc. for C₃₀H₁₈O₁₀.

Hinokiflavone (7): M.P. 362-363°C. UV λ_{\max} (MeOH) 419, 338 and 270 nm. IR ν_{\max} (KBr): 3500, 1635, 1610, 1570 and 1170 cm⁻¹. ¹H- and ¹³C-NMR (DMSO-*d*₆): Tables 3, 4. ESIMS: 539 (100, [M + 1]⁺), 561 (64, [M + Na]⁺), 577 (28, [M + K]⁺), calc. for C₃₀H₁₈O₁₀.

Sucrose (8). M.P. 183-184 °C. $[\alpha]_D = +68.0$ ($c = 0.25$, H₂O). ESIMS: 365 (100, [M + Na]⁺), 381 (53, [M + K]⁺), calc. for C₁₂H₂₂O₁₁.

Hypolaetin-7-O- β -xylopyranoside (9). M.P. 199-201°C. UV λ_{\max} (MeOH) 340, 301 and 275 nm; (NaOMe) 392, 335 and 265 nm; (AlCl₃) 363, 316 and 278 nm;

(AlCl₃/HCl) 356, 314 and 281 nm; (NaOAc) 400, 330 and 273 nm. ESIMS: 435 (100, [M + 1]⁺), 457 (67, [M + Na]⁺), calc. for C₂₀H₁₈O₁₁.

(-)-**Catechin (10)**. M.P. 155-156 °C; $[\alpha]_D = -6.0$ ($c = 0.027$, MeOH). UV λ_{\max} (MeOH) 280 nm. ESIMS: 291 (100, [M + 1]⁺), calc. for C₁₅H₁₄O₆.

Animals

Wistar albino rats (150-200 g) roughly the same age (8-10 weeks), obtained from the Experimental Animal Care Center, College of Pharmacy, King Saud University, Riyadh were used. The animals were housed under constant temperature (22 ± 2°C), humidity (55%) and light/dark conditions (12/12 h). They were provided with Purina chow and free access to drinking water *ad libitum* (Alqasoumi *et al.*, 2009). The experiments and procedures used in this study were approved by the Ethical Committee of the College of Pharmacy, King Saud University and the Ethical Committee of the College of Pharmacy, Salman Bin Abdulaziz University.

Chemicals

Silymarin (Sigma-Aldrich, St. Louis, MO, USA).

Hepatoprotective activity

Male Wistar rats were divided into seven groups six animals each. *Group I* received normal saline and was kept as a control group. *Groups II- IX* received 0.125 ml of CCl₄ in liquid paraffin (1:1) per 100 g body weight intraperitoneally. *Group II* received only CCl₄ treatment. *Group III* was administered silymarin at a dose of 10 mg/kg p.o. (20.7 mmole/ kg). *Groups IV and V* were treated with 250 and 500 mg/kg of fractions or 20.7 mmole/ kg of pure compounds. *Group VI* received 5.6 mg of **6** and 3 mg of sugiol (10.35 mmole/ kg). *Group VII* received 5.6 mg of **7** and 3 mg of sugiol (10.35 mmole/ kg). *Group VIII* received 10 mg of silymarin and 6 mg of sugiol (20.7 mmole/ kg). *Group IX* received 11.2 mg of **7** and 6 mg of sugiol (20.7 mmole/kg). Drug treatment was started 5 days prior to CCl₄ administration and continued till the end of the experiment. After 48 h, following CCl₄ administration the animals were sacrificed using ether anesthesia. Blood samples were collected by heart puncture and the serum was separated for evaluating the biochemical parameters. The liver was immediately removed and a small piece was fixed in 10% formalin for histopathological assessment.

Determination of enzyme levels

The biochemical parameters such as serum glutamate oxaloacetate transaminase (SGOT), serum glutamate pyruvate transaminase (SGPT), alkaline phosphatase (ALP) and total bilirubin were estimated by reported methods (Edwards and Bouchier, 1991). The enzyme activities were measured using Reflotron[®] diagnostic strips (Roche, Basel, Switzerland) and were read on a Reflotron[®] Plus instrument (Roche).

Estimation of NP-SH

Non-protein sulphhydryl groups (NP-SH) were quantified according to the Sedlak and Lindsay method (Sedlak and Lindsay, 1968). The liver was cooled in a beaker immersed in an ice bath. 200/mg of liver were homogenized in 8/ml of 0.02 M ethylenediaminetetraacetic acid (EDTA). Aliquots of 5ml of the homogenate were mixed in 15 ml test tubes with 4 ml of distilled water and 1/ml of 50% trichloroacetic acid (TCA). The tubes were shaken intermittently for 10-15 min and centrifuged for 15 min at approximately 3000 rpm to precipitate the protein. 2/ml of the supernatant were mixed with 4/ml of 0.4 M Tris buffer, pH 8.9 and 0.1 ml of 0.01 M DTNB [5, 5'-dithio-bis-(2-nitrobenzoic acid)] added and the sample was shaken. The absorbance was measured spectrophotometrically within 5 min of addition of DTNB at 412 nm against a reagent blank with no homogenate.

STATISTICAL ANALYSIS

For each set of experiments where two or more than two groups were compared, an analysis of variance (ANOVA) test were used to determine the significance of the differences. Differences between the control and CCl₄-treated group were compared for significance using Dunnette test for non paired samples (Woolson and Clarke, 2002). All the values shown are the mean ± S.E.

Histopathology

The fixed livers samples were placed in cassettes and loaded into tissue baskets. They were subjected to dehydration, clearing and infiltration by immersion in different concentrations of ethanol (70- 100%), xylene (3 times, 1hr each) and finally paraffin wax (4 times, 1hr

each). The tissues were then transferred into moulds filled with paraffin wax. After orienting the tissues by hot forceps the moulds were chilled on cold plates and excess wax were trimmed off using a knife. The rotary microtome (Leitz 1512) was used for making thin sections (3µm). The sections were placed onto clean slides that were drained vertically for several minutes before placing them onto a warming table at 37-40°C (Prophet *et al.*, 1994). The slides were then deparaffinized, hydrated and to stained in Mayer's hematoxylin solution for 15 minutes. The slides were then washed in lukewarm running tap water for 15 minutes and placed in distilled water. After they were immersed in 80% ethyl alcohol for one to two minutes then counterstained in eosin-phloxine solution for 2 minutes. The slides were then dehydrated and cleared through two changes each of 95% ethyl alcohol, absolute ethyl alcohol, and xylene (2 minutes each) and finally mounting with resinous medium.

RESULTS

¹H- and ¹³C-NMR data used for the identification of compounds **1- 5** are presented in tables 1 and 2. ¹H- and ¹³C-NMR data of compounds **6, 7** and **9** are presented in tables 3 and 4. Effect of the isolated compounds on serum biochemical parameters using silymarin as standard is presented in table 5. Effects of combination of **6, 7** and silymarin with sugiol on serum biochemical parameters and nonprotein sulphhydryl groups (NP-SH) are presented in tables 6- 9. Histopathological changes of liver cells treated with CCl₄, CCl₄ and silymarin, CCl₄ and compounds **2, 6, 7**, combinations of **7** and silymarin with sugiol comparing with normal liver cells are presented in fig. 1.

Table 1: ¹H-NMR chemical shift values of diterpenes **1-5**

H	* 1	* 2	† 3	† 4	* 5
1	1.10 (brd, 14.0), 1.79 (m)	1.08 (m); 1.86 (m)	1.22 (m), 1.78 (m)	1.06 (m), 1.24 (m)	1.05 (m), 1.78 (m)
2	1.42 (m), 1.76 (m)	1.54 (m); 1.86 (m)	1.44 (m), 1.60 (m)	1.37 (m), 1.40 (m)	1.48 (m), 1.70 (m)
3	1.04 (dd; 4.0; 13.0) 2.16 (brd; 13.0)	1.06 (dd; 4.0; 13.0) 2.17 (brd; 13.0)	1.60 (m), 1.90 (m)	4.02 (dd; 4.5, 11.0)	1.60 (m), 1.70 (m)
5	1.23 (m)		2.44 (dd; 2.5, 13.0)	1.53 (m)	1.88 (m)
6	1.81 (m), 1.88 (m)	1.34 (m)	1.47 (m), 1.70 (m)	1.00 (m), 1.30 (m)	1.64 (m), 1.93 (m)
7	1.85 (m), 2.41 (brd; 9.0)	1.90 (m); 1.98 (m)	4.11 (brs)	1.84 (m), 2.20 (m)	5.24 (s)
9	1.47 (m)	¹ 9.1 (m) 2.42 (dd; 2.5, 8.5)	2.23 (dd; 6.0, 12.5)	1.60 (m)	1.68 (m)
11	1.23 (m), 1.49 (m)		1.44 (m), 1.60 (m)	1.39 (m), 1.43 (m)	1.28 (m), 1.48 (m)
12	1.23 (m), 1.63 (m)	1.54 (m)	1.48 (m), 1.49 (m)	1.15 (m), 1.24 (m)	1.28 (m), 1.41 (m)
14	5.92 (dd; 11.0, 17.5)		5.52 (s)	5.25 (s)	1.84 (m), 1.88 (m)
15	5.08 (dd; 1.0, 11.0) 5.22 (dd; 1.0, 17.5)	1.27 (m), 1.52 (m)	5.82 (dd; 10.5, 17.5)	5.76 (dd; 11.0, 18.0)	5.73 (dd; 10.5, 17.5)
16	1.20 (s)	0.96 (m) 1.45 (m)	4.93 (d, 10.5) 5.01 (d, 17.5)	4.63 (d, 11.0) 4.68 (d, 18.0)	4.79 (d, 10.5) 4.85 (d, 17.5)
17	4.50 (s), 4.85 (s)	1.54 (m)	1.07 (s)	1.05 (s)	0.79 (s)
18	1.16 (s)	1.34 (m); 1.65 (m)	—	—	—
19	—	3.64–3.72 (m)	1.19 (s)	1.14 (s)	1.19 (s)
20	0.51 (s)	0.92 (d; 6.5)	0.86 (s)	0.87 (s)	0.83 (s)

*CDCl₃ and †CD₃OD, δ in ppm, J in Hz.

Table 2: ^{13}C -NMR chemical shift values of diterpenes **1-5**

Carbon	* 1	* 2	† 3	† 4	* 5
1	39.1	39.2	39.5	38.3	38.8
2	19.9	19.9	19.3	19.8	17.9
3	37.9	38.0	38.1	76.4	37.0
4	44.2	44.2	48.0	54.7	46.3
5	56.4	56.4	43.3	51.8	45.0
6	26.0	26.1	33.2	25.5	25.2
7	38.7	38.8	73.8	36.6	121.0
8	148.0	148.3	139.8	137.6	135.6
9	56.5	56.7	47.4	51.5	52.0
10	40.6	40.6	39.1	38.6	35.0
11	17.9	21.2	19.3	27.9	20.0
12	41.3	36.4	35.3	35.7	36.1
13	73.7	30.3	38.5	38.5	36.8
14	145.0	39.6	135.1	130.4	46.1
15	111.7	61.3	149.3	149.9	150.3
16	28.0	19.8	111.3	110.8	109.2
17	106.5	106.4	26.3	26.6	21.5
18	29.0	29.0	182.5	181.3	184.9
19	183.5	183.4	17.5	11.7	17.1
20	12.7	12.8	15.1	15.7	15.2

*CDCl₃ and †CD₃OD, δ in ppm, J in Hz.**Table 3:** ^1H -NMR chemical shift values of Compounds **6, 7** and **9** (DMSO- d_6)

Position	6-I	6-II	7-I	7-II	9
3	6.80 (s)	6.80 (s)	6.85 (s)	6.84 (s)	6.63 (s)
5	13.1 (s) (OH)	13.1 (s) (OH)	12.89 (s) (OH)	13.22 (s) (OH)	—
6	6.48 (s)	6.48 (s)	6.20 (d; 2.0)		6.48 (s)
7	10.8 (s) (OH)	10.8 (s) (OH)	11.30 (s) (OH)	10.86 (s) (OH)	—
8			6.48 (d; 2.0)	6.74 (s)	—
2'	7.51 (dd; 2.0, 8.5)	7.51 (dd; 2.0, 8.5)	8.01 (d; 8.5)	7.96 (d; 8.5)	7.40 (s)
3'	6.76 (dd; 2.0, 8.5)	6.76 (dd; 2.0, 8.5)	7.05 (d; 8.5)	6.95 (d; 8.5)	—
4'	10.3 (s) (OH)	10.3 (s) (OH)		10.39 (s) (OH)	—
5'	6.76 (dd; 2.0, 8.5)	6.76 (dd; 2.0, 8.5)	7.05 (d; 8.5)	6.95 (d; 8.5)	6.81 (brd; 8.0)
6'	7.51 (dd; 2.0, 8.5)	7.51 (dd; 2.0, 8.5)	8.01 (d; 8.5)	7.96 (d; 8.5)	7.39 (brd; 8.0)

Xyl: 4.86 (d; $J=7.5$, (H-1')), 3.29 (m, H-2'), 3.20 (m, H-3'), 3.31 (m, H-4'), 3.29 (m); 3.73 (m, H-5').**Table 4:** ^{13}C -NMR chemical shift values of Compounds **6, 7** and **9** (DMSO- d_6)

Position	6-I	6-II	7-I	7-II	9
2	163.0	163.0	163.1	164.2	164.2
3	103.1	103.1	103.9	102.6	102.7
4	182.5	182.5	181.7	182.1	182.2
5	161.5	161.5	161.3	153.1	152.2
6	99.0	99.0	98.9	124.7	98.5
7	164.1	164.1	164.2	157.3	150.8
8	98.8	98.8	94.0	94.6	127.1
9	155.3	155.3	157.1	153.7	144.5
10	104.2	104.2	103.8	104.1	105.2
1'	121.7	121.7	124.2	121.1	121.6
2'	128.4	128.4	128.3	128.5	113.6
3'	116.3	116.3	115.3	115.9	145.7
4'	161.4	161.4	161.4	160.6	149.8
5'	116.3	116.3	115.3	115.9	116.0
6'	128.4	128.4	128.3	128.5	119.1

Xyl: 101.7 (C-1'), 73.0 (C-2'), 75.6 (C-3'), 69.3 (C-4'), 65.8 (C-5').

Table 5: Effects of compounds **2**, **6** and **7** on serum biochemical parameters

Treatment (n=6)	Biochemical Parameters							
	SGOT (units/l)		SGPT (units/l)		ALP (units/l)		Bilirubin (mg/dl)	
	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease
Normal	104.96±24.27		33.10±10.31		497.00±31.91		0.67±0.12	
CCl ₄ only	415.50 ^a ±23.01*		362.50 ^a ±41.89*		1105.33 ^a ±41.96*		4.22 ^a ±0.25*	
Silymarin+ CCl ₄	176.33 ^b ±31.23*	57.5	106.68 ^b ±22.73*	70.6	569.33 ^b ±29.30*	48.5	1.08 ^b ±0.17*	74.4
2 + CCl ₄	210.83 ^b ±20.44*	49.2	138.78 ^b ±23.51*	61.7	614.00 ^b ±23.07*	44.4	1.61 ^b ±0.31*	61.8
Normal	100.45±15.84		38.45±7.55		427.66±51.14		0.70±0.11	
CCl ₄ only	423.33 ^a ±24.51*		367.00 ^a ±27.12*		1109.50 ^a ±33.51*		4.00 ^a ±0.23*	
Silymarin+ CCl ₄	154.00 ^b ±27.61*	63.6	123.85 ^b ±29.34*	66.2	482.16 ^b ±41.00*	56.5	1.15 ^b ±0.19*	71.2
6 + CCl ₄	171.66 ^b ±32.58*	59.4	124.74 ^b ±24.58*	66.0	508.66 ^b ±23.23*	54.1	1.15 ^b ±0.23*	71.2
Normal	102.81±18.90		40.96±10.57		596.16±38.70		0.71±0.10	
CCl ₄ only	594.16 ^a ±28.30*		474.66 ^a ±36.33*		1206.66 ^a ±36.32*		4.05 ^a ±0.37*	
Silymarin+ CCl ₄	203.50 ^b ±26.17*	69.2	121.26 ^b ±18.84*	74.4	757.66 ^b ±33.42*	37.2	1.36 ^b ±0.19*	66.41
7 + CCl ₄	183.16 ^b ±29.19*	65.7	132.66 ^b ±21.74*	72.0	765.70 ^b ±34.00*	36.5	1.56 ^b ±0.30*	61.5

*p<0.001, ^aas compared with the normal saline (control) group; ^b as compared with the CCl₄ only group

Table 6: Effects of combination of **6**, **7** and sugiol at 10.35 mmole/kg on serum biochemical parameters

Treatment (n=6)	Biochemical Parameters							
	SGOT (units/l)		SGPT (units/l)		ALP (units/l)		Bilirubin (mg/dl)	
	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease
Normal	97.26±11.31		28.85±5.86		385.83±23.04		0.58±0.05	
CCl ₄ only	428.83 ^a ±34.38 **		354.33 ^a ±30.44**		914.33 ^a ±41.79**		4.05 ^a ±0.17**	
Silymarin + CCl ₄	140.83 ^a ±13.64*	67.15	106.18 ^a ±13.85**	70.03	467.83 ^a ±29.26**	48.83	1.07 ^a ±0.13**	73.58
6 + Sugiol +CCl ₄	264.00 ^b ±23.83*	38.43	207.5 ^b ±18.10*	41.43	681.50 ^b ±27.63**	25.46	2.37 ^b ±0.25**	41.48
7 + Sugiol +CCl ₄	196.83 ^b ±22.71**	54.10	165.5 ^b ±21.36**	53.29	595.5 ^b ±21.87**	34.87	1.98 ^b ±0.20**	51.11

*p<0.01, **p<0.001, ^a as compared with the normal saline (control) group; ^b as compared with the CCl₄ only group

Table 7: Effects of combination of Silymarin, **7** and sugiol at 20.7 mmole/kg on serum biochemical parameters

Treatment (n=6)	Biochemical Parameters							
	SGOT (units/l)		SGPT (units/l)		ALP (units/l)		Bilirubin (mg/dl)	
	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease	Mean ±S.E.	% Decrease
Normal	85.18±11.22		32.93±6.23		308.6±13.72		0.59±0.05	
CCl ₄ only	398.00 ^a ±21.50		340.16 ^a ±30.96		990.5 ^a ±35.36		3.89 ^a ±0.27	
Silymarin + CCl ₄	122.43 ^a ±14.96*	69.23	122.83 ^a ±10.36*	63.89	426.5 ^a ±25.04*	56.94	1.2 ^a ±0.26*	69.15
Silymarin+ Sugiol+CCl ₄	108.46 ^b ±14.63*	72.74	101.61 ^b ±15.63*	70.12	378.83 ^b ±23.61*	61.75	1.09 ^b ±0.17*	71.97
7 + Sugiol +CCl ₄	123.00 ^b ±13.40*	69.09	115.66 ^b ±15.17*	65.99	403.83 ^b ±23.26*	59.22	1.26 ^b ±0.13*	67.60

*p<0.001, ^a as compared with the normal saline (control) group; ^b as compared with the CCl₄ only group

DISCUSSION

Our previous study of the hepatoprotective effect of the

total extract and fractions obtained by liquid-liquid fractionation of the aerial parts of *J. phoenicea* based on biochemical parameters (serum glutamate oxaloacetate

Table 8: Effects of combination of **6**, **7** and sugiol at 10.35 mmole/kg on the level of NP-SH groups

Treatment (n=6)	NP-SH		
	Mean \pm S.E. ($\mu\text{mol/gm}$ wet weight tissue)	% Increase with reference to CCl_4 -treated liver	% effectiveness for reversing to normal
Normal (control)	4.25 \pm 0.51		
CCl_4 only (toxicity control)	1.58 \pm 0.12 *		
Silymarin + CCl_4	3.60 \pm 0.58 *	128.66	84.71
6 + Sugiol+ CCl_4	2.16 \pm 0.29 *	36.70	50.83
7 + Sugiol+ CCl_4	2.30 \pm 0.23 *	45.56	54.12

*) $p < 0.001$, as compared with the CCl_4 only group

Table 9: Effects of combination of Silymarin, **7** and sugiol at 20.7 mmole/kg on the level of NP-SH groups

Treatment (n=6)	NP-SH		
	Mean \pm S.E. ($\mu\text{mol/gm}$ wet weight tissue)	% Increase with reference to CCl_4 -treated liver	% effectiveness for reversing to normal
Normal (control)	4.86 \pm 0.20		
CCl_4 only (toxicity control)	1.57 \pm 0.18 *		
Silymarin + CCl_4	4.26 \pm 0.47 *	171.33	87.66
Silymarin + Sugiol+ CCl_4	4.29 \pm 0.45 *	173.24	88.28
7 + Sugiol+ CCl_4	3.85 \pm 0.25 *	145.24	79.22

*) $p < 0.001$, as compared with the CCl_4 only group

transaminase (SGOT), serum glutamate pyruvate transaminase (SGPT), alkaline phosphatase (ALP) and bilirubin), hepatic nonprotein sulfhydryl group (NP-SH) and histopathological examination showed promising results. Toxicity study were also preformed. According to the obtained results, the methanol sub-fraction is the most safe and active (Alqasoumi *et al.*, 2009). Based on these finding we carried out a detailed biologically directed phytochemical study.

Compounds **8** and **10** were identified as sucrose (Popov *et al.*, 2006) and (-)-catechin (Lu and Foo, 1999) by comparison of their data with the literature as well as direct comparison with authentic samples.

The ESIMS of compound **1** and **2** showed $\text{M}^+ + \text{Na}$ at 343 and 345 m/z consistent with molecular formulae $\text{C}_{20}\text{H}_{32}\text{O}_3$ and $\text{C}_{20}\text{H}_{34}\text{O}_3$ respectively. Both ^1H - and ^{13}C -NMR (tables 1, 2) of **1** and **2** showed two proton singlets at d_{H} 4.50, 4.85 and 4.51, 4.85 respectively assigned for H-17. Both compounds showed an 18- CH_3 singlet (d_{H} 1.16, d_{C} 29.0)(d_{H} 1.25, d_{C} 29.0); 19-COOH (d_{C} 183.5)(d_{C} 183.4); 20- CH_3 singlet (d_{H} 0.51, d_{C} 12.7) (d_{H} 0.61, d_{C} 12.8) respectively. These data are typical for laban-8(17)-ene skeleton (Su *et al.*, 1994). ^1H - and ^{13}C -NMR (tables 1, 2) of **1** showed another monosubstituted double bond (CH at d_{H} 5.92, dd, $J = 11.0, 17.5$ Hz and d_{C} 145.0; CH_2 at d_{H} 5.08 dd, $J = 1.0, 11.0$ Hz, d_{H} 5.22 dd, $J = 1.0, 17.5$ Hz and d_{C} 111.71) assigned for C-14 and C-15. The methyl singlet (d_{H} 1.20, d_{C} 28.0) was assigned to C-16 on an oxygenated fully substituted C-13 at d_{C} 73.7 as indicated from HMBC experiments. The data of **1** was identical with those reported for 13-epicupressic acid (Su *et al.*,

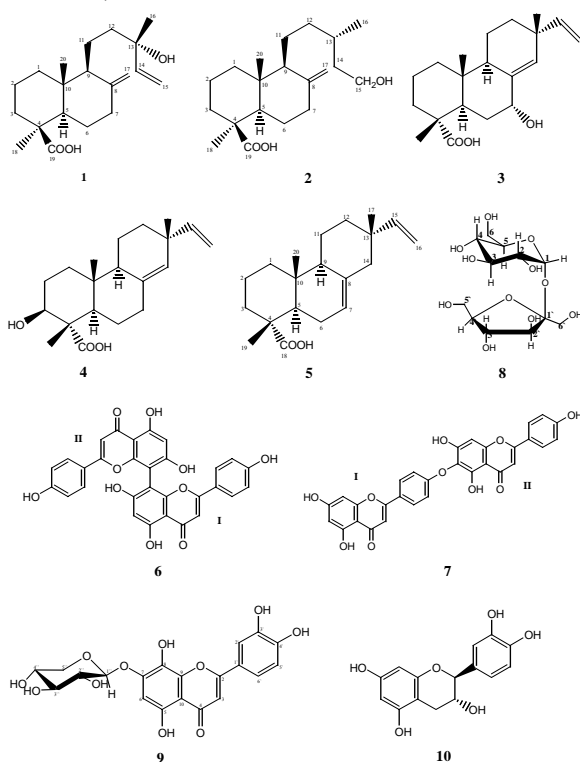
1994). MS, ^1H - and ^{13}C -NMR data of **2** indicated the lack of the second double bond. The appearance of the C-16 methyl as doublet at d_{H} 0.92 ($J = 6.5$ Hz) and the appearance of CH_2OH (d_{H} 3.64–3.72 (m), d_{C} 61.3) indicated a C-15 hydroxyl in **2**. The data of **2** was consistent with those reported for imbricatolic acid (Wenkert and Buckwalter, 1972).

Both **3** and **4** (table 1, 2) have a pimaradiene skeleton with a C-8, C-14 double bond (d_{C} 139.8, 135.1 and d_{C} 137.6, 130.4 respectively) (Su *et al.*, 1994). They also showed same EISMS $\text{M}^+ + \text{Na}$ at 341 m/z consistent with the molecular formula $\text{C}_{20}\text{H}_{30}\text{O}_3$. In each compound there were an oxygenated CH (d_{H} 4.11 brs, d_{C} 73.8; d_{H} 4.02 dd, $J = 4.5, 11.0$ Hz, d_{C} 76.4 respectively). The locations of the hydroxyl groups were assigned to C-7 and C-3 in **3** and **4** respectively based on the HMBC correlations. Comparison of their data with the literature identified **3** as 7 α -Hydroxysandaracopimaric acid and **4** as 3 β -Hydroxysandaracopimaric acid, respectively (Muhammad *et al.*, 1992; Su *et al.*, 1994; Wenkert and Buckwalter, 1972).

The molecular formula of **5** ($\text{C}_{20}\text{H}_{30}\text{O}_2$, ESIMS $\text{M}^+ + \text{Na}$ at 325 m/z) as well as ^1H - and ^{13}C -NMR (tables 1,2) indicated a pimaradiene skeleton with one oxygen less than **3** and **4** (Wenkert and Buckwalter, 1972). In the ^{13}C -NMR spectrum the two carbon signals at d_{C} 121.0 and d_{C} 135.6 were assigned for C-7 and C-8 respectively based on HMBC experiment and the compound was identified as isopimaric acid (Wenkert and Buckwalter, 1972; Mihashi *et al.*, 1969; Topcu and Ulubelen, 1990).

UV data of **6** indicated a 5, 7, 4'-trihydroxy flavone

(Mabry *et al.*, 1970). $^1\text{H-NMR}$ (tables 3) showed one singlet for ring A at d_{H} 6.48, one singlet for H-3 at d_{H} 6.80 and two doublets each integrated for two protons at d_{H} 6.76 and 7.51 assigned for 4'-monosubstituted ring B. $^{13}\text{C-NMR}$ data (tables 4) indicated that C-8 appeared as fully substituted at d_{C} 98.8 bears non-oxygenated substituent (Mabry *et al.*, 1970; Agrawal, 1989). However, ESIMS showed peaks at m/z 539 $[\text{M} + 1]^+$, 561 $[\text{M} + \text{Na}]^+$ and 577 $[\text{M} + \text{K}]^+$ calc. for $\text{C}_{30}\text{H}_{18}\text{O}_{10}$ indicating a dimeric structure. Similarly, the data of **7** indicated 5, 7, 4'-trihydroxy flavone. ESIMS showed m/z 539 $[\text{M} + 1]^+$, 561 $[\text{M} + \text{Na}]^+$ and 577 $[\text{M} + \text{K}]^+$ calc. for $\text{C}_{30}\text{H}_{18}\text{O}_{10}$. $^1\text{H-}$ and $^{13}\text{C-NMR}$ showed two H-3 protons at d_{H} 6.84 and 6.85. One part of the compound showed two doublets ($J=2.0$ Hz) at d_{H} 6.20 and 6.48 assigned for H-6 and H-8 respectively. The other part of the compound showed only one singlet for ring A at 6.74 assigned for H-8. Comparison with literature data enable the identification of **6** as cupressuflavone and **7** as hinokiflavone (Markham *et al.*, 1987; Kang *et al.*, 2005; Williams *et al.*, 1987; Chari *et al.*, 1977).



The UV data of **9** pointed out to a flavone derivative ($\lambda_{\text{max}}^{\text{MeOH}}$ 340 nm) with free 5, 3', 4'-trihydroxy substituents. The lack of any shift with the use of NaOAc indicated a substituted C-7 hydroxyl (Mabry *et al.*, 1970). The $^1\text{H-NMR}$ data of **9** (table 3) showed two singlets at d_{H} 6.63 and 6.48 correlated in an HSQC experiment with two carbon signals at d_{C} 102.7 and 98.5 respectively assigned for H-3 and H-6. Ring B protons appeared as an ABX system at d_{H} 7.40 (two overlapped protons H-2' and H-6') and d_{H} 6.81 (d, $J=8.0$ Hz, H-5'). All the assignments were supported by an HSQC and HMBC experiments.

Both $^1\text{H-}$ and $^{13}\text{C-NMR}$ data showed signals for pentose moiety. This proposal was supported by the ESIMS $\text{M}^+ + \text{Na}$ at m/z 457 consistent with the molecular formula $\text{C}_{20}\text{H}_{18}\text{O}_{11}$. The location of the sugar part at C-7 was further supported by the HMBC correlation between the H-1'' at d_{H} 4.86 (d, $J=7.5$ Hz) and C-7 at d_{C} 150.8. The chemical shifts of the sugar carbons as well as direct comparison with standard sugars after hydrolysis led to the identification of the C-7 moiety as *b-D*-xylose (Agrawal, 1989). Compound **9** was identified as hypolaetin-7-*O*-*b*-xylopyranoside previously isolated from *Juniperus communis* var. *depressa* (Dai *et al.*, 2006; Nakanishi *et al.*, 2004).

Treatment of animals with the hepatotoxic agent carbon tetrachloride resulted in significant increase of transaminases (SGOT and SGPT) and alkaline phosphate levels (ALP) due to hepatocytes damage (Zafar and Ali, 1998). Sever jaundice was reflected by increase level of serum bilirubin (Lin *et al.*, 1997). Silymarin act as an antioxidant by scavenging prooxidant free radicals and by increasing the intracellular concentration of GSH. It also exhibits a regulatory action of cellular membrane permeability and increase in its stability against xenobiotics injury; increasing the synthesis of ribosomal RNA by stimulating DNA polymerase-I, exerting a steroid like regulatory action on DNA transcription and stimulation of protein synthesis and regeneration of liver cells (Alqasoumi and Abdel-Kader, 2012b).

All isolated compounds were tested for their possible hepatoprotective effect at a 20.7 $\mu\text{mol/kg}$ dose. However, compounds **2**, **6** and **7** were the most effective in reducing elevated levels of SGOT, SGPT, ALP and bilirubin (table 5). All values were highly significant ($p < 0.001$). Compound **7** was the most active and closely comparable to silymarin followed by **6**. Histopathological appearance (fig. 1) of liver cells of rats treated with CCl_4 and silymarin showed only minimal portal inflammation. Liver cells of rats treated with CCl_4 and **2** showing a good recovery with absence of necrosis, fatty depositions and only minimal lobular inflammation was observed. Histological appearance of rat liver treated with CCl_4 and **6** showed cytoplasmic vascular degenerations around portal tracts, mild inflammation and foci of lobular inflammation. Liver tissue of rats treated with CCl_4 and **7** showed good recovery with absence of necrosis and fatty depositions. The central vein has minimal portal inflammation. It showed superiority as compared with other groups.

Our previous investigation of *J. procera* showed that the abietane diterpene sugiol was the most effect hepatoprotective compound in the plant (Alqasoumi and Abdel-Kader, 2012a). Compounds **6** and **7** are similar in nature to silymarin and most likely act by the same mechanism; however, sugiol has different nature and may

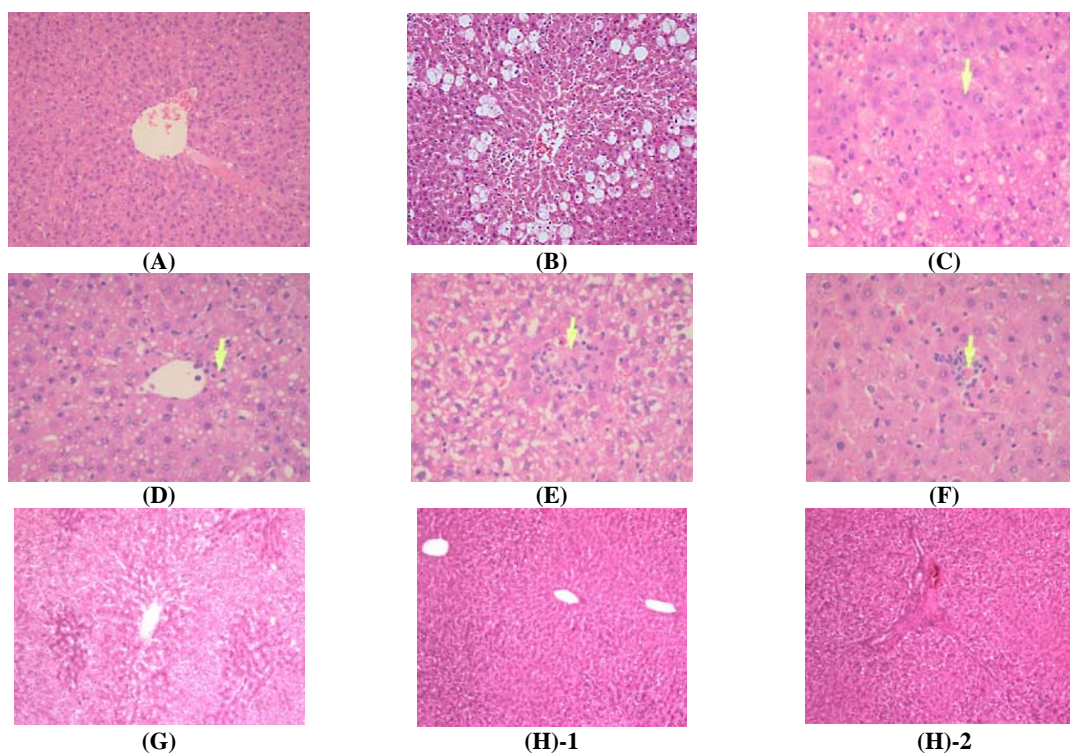


Fig. 1: Histopathological appearance of liver cells; (A) normal cells; (B) liver cells of rats treated with CCl_4 showed centrilobular necrosis and extensive fatty change was observed on the midzonal or entire lobe at 24 h after treatment; (C) liver cells of rats treated with CCl_4 and silymarin showed no necrosis or fatty deposition but had only minimal portal inflammation; (D) liver cells of rats treated with CCl_4 and **2** showing a good recovery with absence of necrosis and fatty depositions and indicating a potent hepatoprotective activity. Only minimal lobular inflammation was found; (E) Histological appearance of rat liver treated with CCl_4 and **6** showed a significant recovery except cytoplasmic vascular degenerations around portal tracts, mild inflammation and foci of lobular inflammation; (F) liver tissue of rats with CCl_4 and **7** showed good recovery with absence of necrosis and fatty depositions. The central vein has minimal portal inflammation; (G) liver cells of rats treated with CCl_4 and combination of silymarin and sugiol showed mild central vacuolar degeneration, fatty change and hydropic degeneration (H) liver tissue of rats with CCl_4 and combination of **7** and sugiol showed normal lobular pattern with only mild congestion with no significant inflammation.

act by different mechanisms. Consequently, combination of **6**, **7** and silymarin with sugiol were to explore the potential in liver protection. Starting with half doses of **6**, **7** and sugiol (10.35 mmole/ kg) in comparison with silymarin resulted in significant improvement of the enzyme levels and hepatic nonprotein sulfhydryl group (NP-SH) (tables 6 and 8). However, non of these treatments exceeded the protective effect of silymarin. Second combinations tested composed of both **7** and silymarin with sugiol at 20.7 mmole/ kg. The obtained results were highly significant ($p < 0.001$) and both combination exceeded the effect of silymarin alone (tables 7 and 9). Histopathological study (fig. 1) showed that treatment with **7** and sugiol resulted in normal lobular pattern with only mild congestion with no significant inflammation. This appearance was the best compared to normal liver cells.

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REFERENCES

- Aboul-Ela M, El-Shaer N and Abd El-Azim T (2005). Chemical constituents and antihepatotoxic effect of the berries of *Juniperus phoenicea* part II. *Nat. Prod. Sci.*, **11**: 240-247.
- Agrawal PK (1989). Carbon-13 NMR of Flavonoids, Elsevier: Amsterdam, Holland.
- Alqasoumi SI and Abdel-Kader MS (2012a). Terpenoids from *Juniperus procera* with hepatoprotective activity. *Pak. J. Pharm. Sci.*, **25**: 315- 322.
- Alqasoumi, SI and Abdel-Kader MS (2012b). Screening of some traditionally used plants for their hepatoprotective effect. in *Phytochemicals as nutraceuticals- global approaches to their role in nutrition and health*. Rao V (ed), InTech., Rijeka, Coratia. <http://www.intechopen.com/books>
- Alqasoumi SI, El Tahir KEH, AlSheikh AM and Abdel-Kader MS (2009). Hepatoprotective Effect and Safety Studies of *Juniperus phoenicea*. *Alex. J. Pharm. Sci.*,

- 23: 81-88.
- Ali SA, Rizk MZ, Ibrahim NA, Abdallah MS, Sharara HM and Moustafa MM (2010). Protective role of *Juniperus phoenicea* and *Cupressus sempervirens* against CCl₄. *World J. Gastrointest. Pharmacol. Ther.*, **1**: 123-131.
- Cairnes DA, Ekundayo O and Kingston DGI (1980). Plant anticancer agents. X. Lignans from *Juniperus phoenicea*. *J. Nat. Prod.*, **43**: 495-497.
- Chari VM, Ilyas M, Wagner H, Neszmelyi A, Chen F-C, Chen L-K, Lin Y-C and Lin Y-M (1977). ¹³C-NMR spectroscopy of biflavanoids. *Phytochemistry*, **16**: 1273-1278.
- Chaudhary SA (1999). Flora of the Kingdom of Saudi Arabia, Vol. **1**, Ministry of Agriculture & Water, Riyadh, p.55.
- Comte G, Allais DP, Chulia AJ, Vercauteren J and Bosso C (1996a). Phoenicoside, the first natural bis-furanone propane derivative from *Juniperus phoenicea* L. *Tetrahedron Lett.*, **37**: 2955-2958.
- Comte G, Allais DP, Chulia AJ, Vercauteren J and Delage C (1996b). Two furanone glucoside derivatives from *Juniperus phoenicea*. *Phytochemistry*, **41**: 1329-1332.
- Comte G, Chulia AJ, Vercauteren J and Allais DP (1996c). Phenylpropane glycosides from *Juniperus phoenicea*. *Planta Med.*, **62**: 88-89.
- Comte G, Vercauteren J, Chulia AJ, Allais DP and Delage C (1997). Phenylpropanoids from leaves of *Juniperus phoenicea*. *Phytochemistry*, **45**: 1679-1682.
- Dai Z, Ma S-C, Wang G-L., Wang F and Lin R-C (2006). A new glucoside from *Selaginella sinensis*. *J. Asian Nat. Prod. Res.*, **8**: 529-533.
- Edwards CRW and Bouchier IAD (1991). Davidson's Principles & Practice Medicine, Churchill Livingstone Press, UK, p. 492.
- Fatma W, Taufeeq HM, Shaida WA and Rahman W (1979). Biflavanoids from *Juniperus macropoda* Boiss and *Juniperus phoenicea* Linn. (Cupressaceae). *Indian J. Chem. B Org.*, **17**: 193-194.
- Hussein SAM, Merfort I and Nawwar MAM (2003). Lignans and other phenylpropanoids from leaves of *Juniperus phoenicea*. *J.S.C.S.*, **7**: 105-110.
- Kang SS, Lee J Y, Choi YK, Song SS, Kim JS, Jeon SJ, Han YN, Son KH and Han BH (2005). Neuroprotective effects of naturally occurring biflavonoids. *Bioorg. & Med. Chem. Lett.*, **15**: 3588-3591.
- Lin CC, Shieh DE and Yen MN (1997). Hepatoprotective effect of fractions Ban-zhi-lian of experimental liver injuries in rats. *J. Ethnopharmacol.*, **56**: 193-200.
- Lu Y and Foo LY (1999). The polyphenol constituents of grape pomace. *Food Chem.*, **65**: 1-8.
- Maatooq GT, El-Sharkawy SH, Afifi MS and Rosazza JPN (1998). Flavonoids from two Cupressaceae plants. *Nat. Prod. Sci.*, **4**: 9-14.
- Mabry TJ, Markham KR and Thomas MB (1970). The Systematic Identification of Flavonoids, Springer-Verlag, New York.
- Markham KR, Sheppard C and Geiger H (1987). ¹³C-NMR studies of some naturally occurring amentoflavone and hinokiflavone biflavonoids. *Phytochemistry*, **26**: 3335-3337.
- Mihashi S, Yanagisawa I, Tanaka O and Shibata S (1969). Further study on the diterpenes of *Aralia* spp. *Tetrahedron Lett.*, **21**: 1683-1686.
- Mossa JS, Muhammad I, El-Ferally FS and Hufford CD (1992). 3b, 12-Dihydroxyabieta-8,11,13-Triene-1-one and other constituents From *Juniperus Excelsa* Leaves. *Phytochemistry*, **31**: 2789-2863.
- Muhammad I, Mossa JS, Al-Yahya MA, Ramadan AF and El-Ferally FS (1995). Further antibacterial diterpenes from the bark and leaves of *Juniperus procera* Hochst. ex Endl. *Phytother. Res.*, **9**: 584-588.
- Muhammad I, Mossa JS and El-Ferally FS (1992). Antibacterial diterpenes from the leaves and seeds of *Juniperus excelsa*. *Phytother. Res.*, **6**: 261-264.
- Muhammad I, Mossa JS and El-Ferally FS (1996). Additional antibacterial diterpenes from the bark of *Juniperus procera*. *Phytother. Res.*, **10**: 604-607.
- Nakanishi T, Iida N, Inatomi Y, Murata H, Inada A, Murata J, Lang FA, Iinuma M and Tanaka T (2004). Neolignan and flavonoid glycosides in *Juniperus communis* var. *depressa*. *Phytochemistry*, **65**: 207-213.
- Popov KI, Sultanova N, Rönkkömäki H, Hannu-Kuure M, Jalonen J, Lajunen LHJ, Bugaenko FI and Tuzhilkin VI (2006). ¹³C NMR and electrospray ionization mass spectrometric study of sucrose aqueous solutions at high pH: NMR measurement of sucrose dissociation constant. *Food Chem.*, **96**: 248-253.
- Prophet EP, Mills B, Arrington JB and Sobin LH (1994). Laboratory Methods in Histology, 2nd Ed. American Registry of Pathology, Washington, D.C.
- Sedlak J and Lindsay RH (1968). Estimation of total, protein-bound, and nonprotein sulfhydryl groups in tissue with Ellman's reagent. *Anal. Biochem.*, **25**: 192-205.
- Su W-C, Fang J-M and Cheng Y-S (1994). Labdanes from *Cryptomeria japonica*. *Phytochemistry*, **37**: 1109-1114.
- Topcu G and Ulubelen A (1990). Diterpenes from *Salvia wiedemannii*. *Phytochemistry*, **29**: 2346-2348.
- Wenkert E and Buckwalter BL (1972). Carbon-13 nuclear magnetic resonance spectroscopy of naturally occurring substances. X. Pimaradienes. *J. Am. Chem. Soc.*, **94**: 4367-4369.
- Williams CA, Harborne JB and Tomas-Barberan FA (1987). Biflavonoids in the primitive monocots *Isophysis tasmanica* and *Xerophyta plicata*. *Phytochemistry*, **26**: 2553-2555.
- Woolson RF and Clarke WR (2002). Statistical Methods for the Analysis of Biomedical Data, 2nd Ed. John Wiley & Sons, Inc., New York.
- Zafar R and Ali SM (1998). Anti-hepatotoxic effects of root and root callus extracts of *Cichorium intybus* L. J. *Ethnopharmacol.*, **63**: 227-231.